SC10F Exam Review

Atoms and Elements

1. Match the words on the left with the definitions on the right.

<u>_</u>	matter	(A)	Any two or more atoms bonded together.
<u></u>	mass	(B)	A change in the form or appearance of a substance.
<u>M</u>	weight	(C)	Anything that occupies space.
<u> </u>	atom	(D)	A pure substance that can be broken down by chemical changes.
<u> </u>	molecule	(E)	Two or more substances that are together and can be separated by physical changes.
<u>D</u>	compound	(F)	A mixture that is the same throughout.
<u> </u>	mixture	(G)	A mixture with visible components.
<u> </u>	homogeneous mixture	(H)	The smallest particle of an element that has the properties of that element.
<u>_</u>	heterogeneous mixture	(I)	When two or more substances join to form new substances with new chemical properties.
<u>_</u>	physical property	(J)	The amount of matter in an object.
<u></u>	chemical property	(K)	The ability (or inability) to change from one type of matter into another type.
<u></u>	physical change	(L)	A characteristic of matter that is not associated with a change in its chemical composition.
<u> </u>	chemical change	(M)	The force of gravity acting on an object.

- 2. Indicate if each of the following is a homogeneous mixture, heterogeneous mixture, compound, or element.
 - (a) Water <u>compound</u>
 - (b) Coffee <u>homogeneous mixture</u>
 - (c) Sugar <u>compound</u>
 - (d) Nitrogen <u>element</u>
 - (e) Buttered popcorn <u>homogeneous mixture</u>
- 3. Indicate if each of these is a physical or chemical property.
 - (a) Boiling point <u>physical</u>
 - (b) Acidity <u>chemical</u>
 - (c) Color <u>physical</u>
 - (d) Flammability <u>chemical</u>
 - (e) Hardness <u>physical</u>

4. Indicate if each of the following is a chemical change or a physical change.

(a) Burning wood	<u>chemical</u>
(b) Tearing a piece of paper in half	<u>physical</u>
(c) Adding sugar to a glass of water	<u>physical</u>
(d) Baking a cake	<u>chemical</u>
(e) Boiling water	<u>physical</u>

5. Complete the following table.

Particle	Symbol	Charge	Location
proton	p^+	positive	nucleus
neutron	n0	neutral	nucleus
electron	е-	negative	clouds surrounding nucleus

6. Complete the following chart.

Element name	Element Symbol	Atomic number	Mass number	Number of protons	Number of electrons	Number of neutrons	Family name
Potassium	K	19	39	19	19	20	Alkali metals
Neon	Ne	10	20	10	10	10	Noble gases
Fluorine	F	9	19	9	9	10	Halogens

- 7. Draw a Bohr diagram of each of the following elements.
 - (a) Sulfur (b) Magnesium $\begin{pmatrix} 16 & p^+ \\ 16 & n^0 \end{pmatrix}$

8. List the properties of metals and non-metals.

Metals	Non-metals
shiny	dull
malleable	brittle
ductile	poor conductors of heat and electricity
good conductors of heat and electricity	

- 9. List the name and quantity of each element in the following compounds.
 - (a) NaNO₃
 - Na Sodium = 1N - Nitrogen = 1O - Oxygen = 3
 - (b) $Mg(OH)_2$

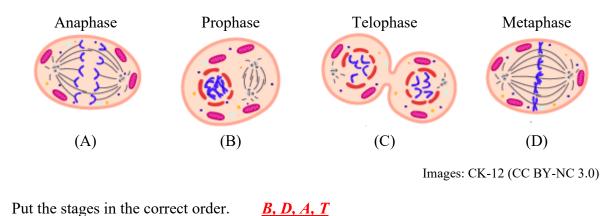
Mg - Magnesium = 1O - Oxygen = 2H - Hydrogen = 2

Reproduction

10. Compare and contrast mitosis and meiosis by completing this table. Use the word phrases below.

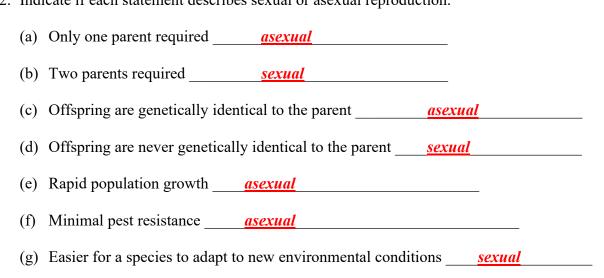
Mitosis	Same for Both	Meiosis
Creates 2 diploid cells	A type of cell division	Creates 4 haploid cells
One stage of division	Creates new cells	Two stages of division
Body cells divide	Replicates DNA	Sex cells divide
No genetic diversity		Creates genetic diversity
Identical to parent cell		Different than parent cell

Creates 4 haploid cells Creates 2 diploid cells A type of cell division One stage of division Two stages of division Creates new cells Body cells divide Sex cells divide Creates genetic diversity No genetic diversity Identical to parent cell Different than parent cell Replicates DNA



11. The following pictures represents cells in the various stages of mitosis.

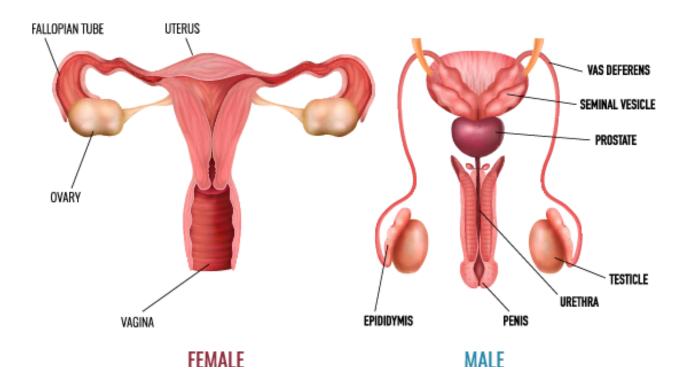
12. Indicate if each statement describes sexual or asexual reproduction.



13. Indicate the type of asexual reproduction being described. A list is provided following the descriptions.

(a) <u>Fragmentati</u>	ion	-	rows to become	t into multiple parts, each e a complete, independent,
(b) <u>Budding</u>			grows out of th into a new indi	e body of the parent, then vidual.
(c) <u>Layering</u>		A stem attached to the plant is bent and covered with soil.		
(d) <u>Vegetative propagation</u>		A plant grows a new shoot which can become a whole new organism.		
(e) <u>Cuttings</u>				taining nodes and bist soil and allowed to
Budding	Fragment	ation	Spores	Vegetative propagation
Cuttings	Grafting		Layering	

14. Label the following diagrams of the human reproductive system.



Credit: macrovector (Adobe Stock Photo)

epididymis	penis	testicle	vagina
fallopian tube	prostate	urethra	vas deferens
ovary	seminal vesicle	uterus	

15. Match the parts of the female and male reproductive systems with the appropriate function.

A.	epidiymus	<u></u>	connect ovaries to the uterus
B.	fallopian tubes	<u><u> </u></u>	produce eggs and secrete estrogen
C.	ovaries	<u> </u>	produce sperm and secrete testosterone
D.	penis	<u><u> </u></u>	passageway for a baby to leave the mother's body
Е.	prostate gland	<u></u>	secrete substances that become part of semen
F.	testes	<u>D</u>	path for sperm to leave body through the urethra
G.	uterus	<u>A</u>	store sperm until they leave the body
H.	vagina	<u> </u>	transport sperm from the epididymis to the urethra
I.	vas defrens	<u>_</u>	where a fetus grows and develops until birth

16.	Indicate the hormone(s)	that is responsible	for the activity.
10.	maleate the normone(b)	that is responsible	for the detivity.

Hormone(s)		Activity
FSH LH		released from the anterior pituitary
FSH LH		stimulate sperm production and testosterone secretion by the testes
Estrogen Progesterone		regulate a female's ovarian and menstrual cycle
Testosterone		responsible for the secondary sexual characteristics that develop in the male during adolescence
Estrogen		responsible for the secondary sexual characteristics of females
Testosterone		stimulates sperm production and operates as a feedback control to the hypothalamus
FSH LH	Testosterone	Estrogen Progesterone

17. Fill in the blanks with words from the word bank.

The male <u>sperm</u>	_ cell and the female	<u>egg</u>	fuse
together to produce a <u>zygote</u>	that tr	avels down	the fallopian tube to
the <u>uterus</u>	It grows as it travels an	nd becomes a	a blastocyst. The
blastocyst embeds in the lining of	the uterus forming an	<u>embryo</u>	The
embryo begins to grow and becom	e more complex. After a	bout eight w	eeks, it has
developed specialized cells and me	ost organs. At this stage	it is now refe	erred to as a
<u>fetus</u>	·		

	1	C .				
egg	embryo	fetus	sperm	uterus	zygote	

The Nature of Electricity

- 18. Fill in the blanks with words from the word bank below. (Some words will be used more than once and other words will not be used at all).
 - (a) There are two types of charges: <u>positive</u> and <u>negative</u>. Objects with the same charge <u>repel</u> each other and objects with opposite charges
 <u>attract</u> each other. An object becomes positively charged when
 <u>electrons</u> are <u>removed</u>. An object becomes negatively charged when
 - (b) When a positively charged rod is brought near a neutral plastic ball. The charges inside the ball separate. This is called <u>polarization</u>. If the charged rod touches the ball, <u>electrons</u> will move from the <u>ball</u> to the <u>rod</u> and the ball will have a <u>positive</u> charge. This process is known as charging by <u>conduction</u>.
 - (c) Cola (A) (B)

A negatively charged balloon is brough near a soda can as shown in picture (A). The <u>electrons</u> move away from the balloon to the far end of the can. The can is then touched by a hand as shown in picture (B). Some <u>electrons</u> move from the <u>can</u> to the <u>hand</u>. If the hand is removed while the balloon is still present, the can will have a <u>positive</u> charge.

added	can	hand	polarization	removed	
attract	conduction	induction	positive	repel	
ball	electrons	negative	protons	rod	

- 19. Label each of the following as a conductor or an insulator.
 - (a) copper <u>conductor</u>
 - (b) plastic <u>insulator</u>
 - (c) rubber <u>insulator</u>
 - (d) aluminum <u>conductor</u>
- 20. Match the words on the left with the definitions on the right.

<u>D</u> conductor	(A) The rate at which charge flows through a circuit.
<u><u> </u></u>	(B) Energy per unit charge.
<u>A</u> current	(C) Rate at which energy is transformed.
<u>B</u> voltage	(D) A material in which electrons can move freely.
<u><u>C</u> power</u>	(E) A material in which electrons cannot move freely.

21. 100 C of charge flows past a point in a circuit in 2 s. Calculate the current in the wire?

$$I = \frac{Q}{t}$$
$$I = \frac{100}{2} = 50 \text{ A}$$

- 22. 2.5 A of current flow through a wire each second. How much charge flows in the same amount of time?
 - $I = \frac{Q}{t}$ $2.5 = \frac{Q}{1}$ Q = 2.5 C

23. 6 C of electric charge flows through a resistor which uses 12 J of energy. What is the potential difference across the resistor?

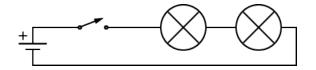
$$V = \frac{E}{Q}$$
$$V = \frac{12}{6} = 2 \text{ V}$$

24. How much energy does each electron ($Q = 1.6 \times 10^{-19}$ C) have in a 9 V battery?

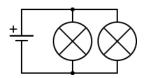
$$V = \frac{E}{Q}$$

9 = $\frac{E}{1.6 \times 10^{-19}}$
E = 1.44 × 10⁻¹⁸ J

- 25. Draw a circuit diagram of an electric circuit with
 - (a) 1 battery, 2 light bulbs and a switch in series.



(b) 1 battery and two light bulbs in parallel.



26. An electric lamp uses a current of 2.5 A when connected to 120 V. Calculate the power the lamp uses?

P = IV $P = 2.5 \times 120 = 300 \text{ W}$

27. Why would it be dangerous to use an extension cord rated for 10 A with a toaster oven that uses 13 A?

When current flows through a wire heat is produced due to the resistance. The cord is only rated for 10 A. If the current is greater than 10 A too much heat could be produced and the cord could catch on fire.

- 28. A TV uses 120 W of power.
 - (a) How much energy in kWh does the TV use if you watch it for 4 hours?

 $Energy = \frac{120}{1000} \times 4 = 0.48 \text{ kWh}$

(b) Electricity costs \$0.10 per kWh. How much would it cost to watch TV 4 hours a day, every day, for the whole year?

 $0.48 \times 365 \times \$0.10 = \17.52

Genetics

- 29. Indicate if the following genotypes are homozygous dominant, homozygous recessive or heterozygous.
 - (a) Dd <u>heterozygous</u>
 - (b) DD <u>homozygous dominant</u>
 - (c) dd <u>homozygous recessive</u>
- 30. In pea plants, purple flowers (P) are dominant to white flowers (p). State the phenotypes for each of the following genotypes.
 - (a) PP <u>purple</u>
 - (b) Pp <u>purple</u>
 - (c) pp <u>white</u>
- 31. In horses, black coat color (B) is dominant to chestnut coat color (b). A heterozygous black coat male is mated with a chestnut coat female.
 - (a) Indicate the genotype of the male and the female horses.

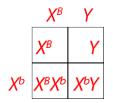
Male:	Bb	Female:	bb

(b) Complete a Punnett square showing the cross between these two horses.

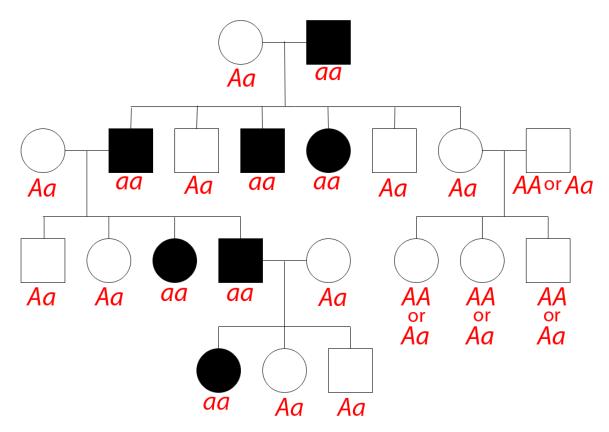
(c) What percentage of the offspring will have black coats?

- 32. Traits controlled by genes located on sex chromosomes are called <u>sex-linked</u> traits.
- 33. Red-green color blindness is a recessive x-linked disorder. Show how it is possible for a normal father to have children that are color blind.

A normal father $(X^B Y)$ can have children that are color blind if the mother has at least one recessive allele (X^b) . She can either be a carrier or be color blind.

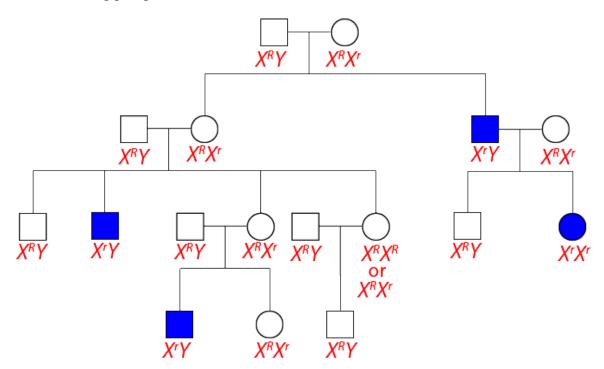


34. The following pedigree tracks an autosomal recessive disorder.



Indicate the genotype of each person in the pedigree. Use "A" for dominant and "a" for recessive.

35. The following pedigree tracks an x-linked recessive disorder.



Indicate the genotype of each person in the pedigree. Use "R" for dominant and "r" for recessive.